The Preparation of Core-Shell CaCO₃ Particles and Its Effect on Mechanical Property of PVC Composites

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ABSTRACT: In this study, a novel mechanochemical route to prepare core-shell structured particles was introduced. XPS, TEM, and dissolving experimental results indicate the formation of [(inorganic particle)/(elastomer)] coreshell structured particles, and several kinds of calcium carbonate (nano-CaCO₃) particles with various interfaces were obtained. The mechanical properties and morphological results indicate that the surface treatment of nano-CaCO₃ particles and the existence of outer elastic layer will strengthen the interfacial interaction between nano-CaCO₃ particles

and PVC matrix, which results in improvement of mechanical properties of PVC/CaCO₃ composites. The theoretical calculations of the interfacial interaction and DMA results confirm these especially when the surface of nano-CaCO₃ particles was treated by MMA and coated in succession by ACR through vibro-milling. © 2006 Wiley Periodicals, Inc. J Appl Polym Sci 102: 1084–1091, 2006

Key words:poly(vinyl chloride); core-shell particles; mechanochemistry; mechanical property

INTRODUCTION

Poly(vinyl chloride) (PVC) is a typical example of brittle thermoplastic and many efforts have been made to improve its fracture resistance for several decades. According to existing reports, modifiers for toughening PVC can be divided into three classes: (1) blending with elastomer, for example, acrylonitrilebutadiene rubber, natural rubber, poly(methyl methacrylate-co-methacrylate) (ACR), chlorinated polyethylene, poly(methacry-*co*-butylene-styrene);¹⁻⁴ (2) blending with short fiber, such as glass fiber, whisker, etc.; 5^{-8} (3) blending with rigid particles, such as poly(methyl methacrylate) (PMMA), calcium carbonate (CaCO₃), and silicon dioxide.⁹ These modifiers show various degrees of toughening PVC but there are respective limitations for them when they are used solely, for example, rigid particles can both toughen and reinforce PVC but only to a certain extent, and blending with elastomers can greatly improve the impact resistance of PVC but often results in a great decline of tensile strength, modulus, and heat resistance. So some researchers used both elastomers and

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rigid particles or short fiber fillers to obtain PVC blends with good performance, and synergic toughening effect was sometimes found.^{10–12} Even then, a modifier owning properties of both elastomer and rigid particles is more advantageous in application. Many studies revealed that core-shell structured particles were a good choice for this aim, because these particles can easily own various properties if the materials of core or shell is designed specially. Until now, core-shell particles used as modifiers are generally obtained from seeded emulsion polymerization and the structure of these core-shell particles are often organic core and organic shell.^{13–16} Although coreshell particles with inorganic core and organic shell have been prepared by sol-gel process or surface grafting and polymerization,^{17,18} preparation of coreshell particles with inorganic core and organic shell prepared in solid is seldom reported and reports about core-shell particles with inorganic core and elastomer shell are also seldom found . Meanwhile, it is well known that solids with activated surfaces can induce various mechanochemical phenomena. Especially, the solid surface has a remarkable reactivity because of the existence of the fresh surfaces and the radical or ion-type active center obtained by vibromilling the solids. It can be used as grafting supports.^{19,20} Through vibro-milling, a homogeneous dispersion of nanoparticles in the matrix can be obtained and aggregation of nanoparticles can be inhibited.^{9,12}

In this study, we prepared core-shell structured particles through vibro-milling method in solid in the presence of nano-CaCO₃ and ACR. The core is nano-

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Sample codes	Treatment mode
CaCO ₃	Untreated nano-CaCO ₃
CaCO ₃ –PMMA	Vibro-milling CaCO ₃ and MMA for 3 h, the weight ratio of
-	CaCO ₃ /MMA is 800/50
CaCO ₃ /ACR	Blending CaCO ₃ and ACR directly
CaCO ₃ -ACR1	Vibro-milling CaCO ₃ and ACR for 1 h
CaCO ₃ -ACR2	Vibro-milling CaCO ₃ and ACR for 2 h
CaCO ₃ -PMMA/ACR	Blending CaCO ₃ -PMMA and ACR directly
CaCO ₃ -PMMA-ACR1	Vibro-milling CaCO ₃ -PMMA and ACR for 1 h
CaCO ₃ –PMMA–ACR2	Vibro-milling CaCO ₃ -PMMA and ACR for 2 h

TABLE ISurface Treatment of Nano-CaCO3

For all treatment modes above, the weight ratio of ACR/CaCO₃ is 20/80, the rotation speed of the vibro-mill is 800 rpm, the ratio of steel ball/nano-CaCO₃ (by mass) is about 13/1, the diameter of the steel ball is between 4 and 12 mm, and vibro-milling is carried out under the protection of nitrogen gas.

 $CaCO_3$ and the shell is ACR. To reinforce the interfacial interaction between nano-CaCO₃ and ACR, $CaCO_3$ is grafted with methyl methacrylate (MMA) by vibro-milling. The characterization of the core-shell structured particle and the effect of core-shell structured particles on mechanical properties and morphology of PVC/nano-CaCO₃ composites were studied.

EXPERIMENTAL

Materials

The poly(vinyl chloride) (PVC) used was SG-7 (supplied by Yibing Tianyuan Stock, China) with a number-average molecular weight of 5.0×10^4 . The PVC was mixed with 4 phr tribasic lead sulfate, 2 phr dibasic lead phosphite, 1 phr stearic acid, and 0.5 phr oxidized polyethylene wax before use. Tribasic lead sulfate, dibasic lead phosphite, stearic acid, and oxidized polyethylene wax were commercial products. The ACR (KM355P) used was supplied by Rohm and Hass China Pte. Nano-CaCO₃ used was supplied by Shandong Shengda nanomaterials, China, with mean particles size ca. 40 nm. MMA used was analytical reagent and washed by alkali liquor and purified by distillation under reduced pressure before use.

Sample preparation

The design of toughening particle needs to take into account a number of general factors. First, it is important to have an outer elastic layer to provide a means of stress transfer from the matrix to the inorganic rigid particles. Second, the interfacial interaction between rigid particles and elastic layer or between elastic layer and the matrix will greatly affect the properties of PVC composites. ACR was chosen as the elastic layer in this study because it is well compatible with PVC. To enhance the interface adhesion between nanoCaCO₃ and ACR, MMA was chosen as surface modified agent.

All kinds of nano-CaCO₃ particles used in this study were listed in Table I and the weight ratio of nano-CaCO₃/ACR is 80/20, the typical mechanochemical preparation of CaCO₃-ACR core-shell structured particles was conducted as follows. Above all, nano-CaCO₃ was dried at 110°C for 15 h to eliminate possible water absorption and then the desired amount of dried nano-CaCO₃ and purified monomer were vibromilled for 3 h in a vibro-milling machine. Third, ACR was added into the vibro-mill to be covibro-milled for 1 or 2 h before the final core-shell structured particles were taken out.

The sample was prepared by blending PVC with a certain amount (0–15 phr) of nano-CaCO₃ particles for a period of 8 min in a twin roll mixer at 170°C and then compression molded at 185°C.

Measurements and characterization

Stress–strain behavior of PVC/nano-CaCO₃ composites was measured on an Instron 4302 tensile machine (Canton, MA) with dumb-bell specimen dimensions of $25 \times 6.5 \times 1 \text{ mm}^3$ and a crosshead speed of 10 mm/min (GB/T 1040–1992). The Izod notched impact strength was measured on a XJ-40A Impact tester (Chengde, China) and conducted according to the regulation mentioned in GB/T 1843–1996. All tests above were carried out at room temperature with six or seven specimens for each sample.

The surface chemical structures of some kinds of CaCO₃ particles were observed by an X-ray photoelectron spectrum (XPS) on these CaCO₃ particles powders. The morphology of a core-shell structured particle was observed by a transmission electron microscope (TEM) (JEM-CX100, Tokyo, Japan) operated at 100 kV, before observation, the core-shell structured particle powder was scattered in alcohol first and



Figure 1 Schematic diagram of longitudinal section of tensile-fractured specimen for SEM observation.

loaded on a special net and dried. On the other hand, a dissolving experiment was carried out to determine whether CaCO₃ particles are coated completely or not. The solvent is hydrochloric acid (0.15N). Some kinds of CaCO₃ particles were weighted (W_1) and dissolved into the hydrochloric acid till there was no gas blowing off, then filtrated, washed, and dried, and weighted insoluble residues (W_2) and percentages (G)of insoluble residues in original particles were calculated as follows: $G = W_2/W_1 \times 100\%$. The phase structure and fractured behavior of the composites were examined by scanning electron microscope (SEM) analysis. For fractured behavior analysis, the tensile fractured surfaces were cut open parallel to tensile direction under liquid nitrogen temperature (see in Fig. 1). For phase structure observation, the impact samples were brittle fractured under liquid nitrogen temperature. The longitudinal sections of the tensile fractured surfaces and the impact fractured surfaces were coated with silver-palladium alloy and examined with a SEM-X650 (Tokyo, Japan) instrument with an accelerating voltage of 20 kV. Two zones of the longitudinal sections were selected for observation as illustrated in Figure 1. The DMA tests were conducted on a DMA-7e (Perkin-Elmer, USA) in a flexural mode (three-point bend) with a frequency of 1 Hz and a heating rate of 2.5°C/min within the range of -50 to 140°C. The specimen dimensions were 20×2.5 $\times 4 \text{ mm}^3$.

RESULTS AND DISCUSSION

Characterization of core-shell structured particles

As shown in Figure 2(a)–2(c), it can be seen that when as a whole $CaCO_3$ particles are treated with different modes, the content of oxygen and calcium of $CaCO_3$ surfaces decrease with the increase of the content of carbon on $CaCO_3$ surface. The peak at 284.8 eV (the C in C—H) is obviously higher than the peak at 289.8 eV (the C in C—O) in the sample of $CaCO_3$ –PMMA, while the later is higher in the pure $CaCO_3$, and compared with $CaCO_3$, the area of C peaks of $CaCO_3$ –PMMA, $CaCO_3$ –ACR1, and $CaCO_3$ –PMMA–ACR1 particles increase distinctly, meanwhile the area of O and Ca peaks decrease obviously, indicating that on the surfaces of these $CaCO_3$ particles the contents of carbon increase, while the contents of oxygen and calcium decrease evidently, especially in the sample of $CaCO_3$ –PMMA–ACR1, the element of calcium almost disappear.



Figure 2 (a) XPS spectra of carbon of different $CaCO_3$ particles; (b) XPS spectra of oxygen of different $CaCO_3$ particles; and (c) XPS spectra of calcium different $CaCO_3$ particles ACR $CaCO_3$ -PMMA-ACR1.





CaCO,-PMMA-ACR1

Figure 3 TEM micrographs of ACR and CaCO₃–PMMA–ACR1 particle.

All information above demonstrates that MMA can be grafted to the surface of CaCO₃ and a large number of CaCO₃ has been coated by ACR through vibromilling. The morphology of ACR and CaCO3-PMMA-ACR1 particles was observed by TEM on ACR and CaCO₃-PMMA-ACR1 particles powder respectively, and the result is shown in Figure 3. It can be seen that there was no core in ACR particle and the surface of CaCO₃-PMMA-ACR1 particle was different from its inner part. From Table II, it can be seen that percentages of insoluble residues in original particles were much higher than the weight ratio of ACR (20%) in these vibro-milled particles, which indicates that many CaCO₃ particles were completely coated by ACR. When the surfaces of $CaCO_3$ particles were grafted by MMA, percentages of insoluble residues in original particles were enhanced. Percentages of insoluble residues in original particles also increased with the increase of vibro-milling time. All these indicated that treatment of surface with MMA and increase of vibro-milling time are beneficial to the formation of core-shell structured particles with complete coat.

Combining results of XPS, TEM, and dissolving experiment, it can be said that a core-shell structured particle was obtained by vibro-milling.

Mechanical properties of PVC/nano-CaCO₃ composites and interfacial interaction between PVC and nano-CaCO₃ particles

The effects of various nano-CaCO₃ particles on mechanical properties of PVC/nano-CaCO₃ composites are shown in Figure 4(a)-4(b). It can be seen that as a whole the impact strength increases with the addition of various nano-CaCO₃ particles, while the tensile strength decreases with the increase of these particles. Meanwhile, comparing $PVC/(CaCO_3/ACR)$ with PVC/(CaCO₃-ACR1) and PVC/(CaCO₃-ACR2), the formation of core-shell structured particles did not result in remarkable change of notched Izod impact strength, although there was a little increase of tensile strength. But comparing $PVC/(CaCO_3-PMMA/ACR)$ with PVC/(CaCO₃–PMMA–ACR1), and PVC/(CaCO₃– PMMA–ACR2), the formation of core-shell structured particles resulted in remarkable increase of notched Izod impact strength and a little increase of tensile strength. All these indicated that the affinity between $CaCO_3$ and ACR is poor; at the absence of PMMA, the core-shell structured particles (CaCO3-ACR1 and CaCO₃-ACR2) were easily destroyed and stress cannot be easily transferred to CaCO₃ particles. In the presence of PMMA, the interfacial interaction between CaCO₃ and ACR was strong so that the core-shell structured particles (CaCO₃–PMMA–ACR1 and CaCO₃–PMMA–ACR2) can easily transfer stress from outer elastic layers to rigid cores before these particles were destroyed, which makes full use of the synergetic effect of elastomer and rigid particle.

Comparing PVC/(CaCO₃/ACR) with PVC/ (CaCO₃–PMMA/ACR), it can be seen that grafting PMMA to surface of CaCO₃ cannot directly toughen PVC/nano-CaCO₃ composites, the reason is that PMMA is more compatible with ACR than PVC, so the grafting PMMA to surface of CaCO₃ cannot lead to improvement of interfacial interaction between CaCO₃ and PVC.

Comparing PVC/(CaCO₃-ACR1) with PVC/ (CaCO₃-ACR2) and PVC/(CaCO₃-PMMA-ACR1)

TABLE IIResults of Dissolving Experiment in Hydrochloric Acid

Sample	G (%)
$CaCO_3$	1.42
$CaCO_3/ACR$	19.41
$CaCO_3-ACR1$	37.87
$CaCO_3-ACR2$	44.95
$CaCO_3-PMMA-ACR1$	42.72
$CaCO_3-PMMA-ACR2$	46.77





Figure 4 (a) Effect of various nano-CaCO₃ particles on notched Izod impact strength of PVC composites and (b) effect of various nano-CaCO₃ particles on tensile strength of PVC composites.

with PVC/(CaCO₃–PMMA–ACR2) respectively, it can be seen that vibro-milling time has a little effect on properties of PVC/nano-CaCO₃ composites during

TABLE III B_{σ} Values from eq. (2) of PVC Composites Filled with
Different CaCO3 Particles

Sample	B_{σ}
PVC/(CaCO ₃ /ACR) PVC/(CaCO ₃ -PMMA/ACR) PVC/(CaCO ₃ -ACR2) PVC/(CaCO ₃ -PMMA-ACR2)	1.35 1.44 2.07 2.42

experimental vibro-milling time range, but the longer the vibro-milling time is, the more remarkable the aggregation of these nano-CaCO₃ particles due to the plastic effect of ACR during vibro-milling.

From all above phenomena, we can draw a conclusion that different interfacial interactions between nano-CaCO₃ particles and PVC or ACR were obtained by the methods used, which results in changes of mechanical properties of $PVC/CaCO_3$ composites.

Meanwhile, the characterization of interfacial interaction gets more and more attention. To confirm that the effect of the interfacial interaction on the mechanical property of the composites, a lot of work has been done and some models have been developed to describe the relation between the measured tensile strength and modulus and the interfacial interaction of polymer composites. Yield stress is proved to be an excellent property to predict interfacial interaction quantitatively in heterogeneous polymer systems. Concerning stress concentration, interface adhesion, and dimensional factor of filler, a model^{21,22} developed previously described the effect of decreasing load-bearing cross section areas and the interfacial interaction on the yield stress of particulate-filled polymeric system. The expression of yield stress takes the equation as follows:

$$\sigma_c = \sigma_m [(1 - \varphi_f) / (1 + 2.5\varphi_f)] \exp(B_\sigma \varphi_f)$$
(1)

where σ_c and σ_m are the yield stresses of the composite and the matrix respectively, φ_f is the volume fraction of the filler in the composite, and B_σ is the parameter characterizing interfacial interaction. The term (1- φ_f)/(1 + 2.5 φ_f) takes into account the decrease of the



Figure 5 Loss modulus of PVC and various PVC/nano-CaCO₃ (100/15) composites.



Figure 6 Loss factor (tan δ) of PVC and various PVC/nano-CaCO₃ (100/15) composites such as PVC/(CaCO₃/ACR), PVC/(CaCO₃-ACR2), PVC/(CaCO₃-PMMA/ACR), and PVC/(CaCO₃-PMMA-ACR2).

effective load-bearing cross section areas owing to the introduction of the filler into the matrix.

To evaluate the interfacial interactions in the composites studied here, eq. (1) could be rearranged as follows:

$$\ln\{(\sigma_c/\sigma_m)[(1+2.5\varphi_f)/(1-\varphi_f)]\} = B_\sigma\varphi_f \qquad (2)$$

According to eq. (2), the parameter B_{σ} was obtained and used for evaluating interfacial interaction quantitatively. The data listed in Table III show that the interfacial interaction of PVC/(CaCO₃–PMMA– ACR2) is the strongest due to the surface treatment of CaCO₃ particles and the formation of core-shell structured particles. The interfacial interaction in PVC/ (CaCO₃/ACR) is comparatively the weakest.

The interfacial interaction of polymer composites can also be evaluated by dynamic mechanical analysis (DMA).^{23,24} It is well accepted that the increase of the interfacial interaction between polymer matrix and filler particles results in increase of the damping force for molecules motion, so the loss modulus and the loss factor (tan δ) of polymer composites would increase. The stronger the interfacial interaction, the higher the loss modulus and the tan δ of polymer composites. In Figures 5 and 6, the DMA results are shown. The tan δ and the loss modulus of various PVC/nano-CaCO₃ composites were higher than those of PVC and the order of the tan δ and the loss modulus are PVC/ $(CaCO_3-PMMA/ACR) > PVC/(CaCO_3/ACR) > PVC$ and $PVC/(CaCO_3-PMMA-ACR) > PVC/(CaCO_3-PMMA-ACR)$ ACR) > PVC. The height and area of glass transition



Figure 7 SEM micrographs of brittle-fractured surfaces of impact samples of PVC/CaCO₃ (100/15)composites: (a) PVC/(CaCO₃/ACR), (b) PVC/(CaCO₃_ACR2), (c) PVC/(CaCO₃-PMMA/ACR), and (d) PVC/(CaCO₃-PMMA-ACR2).



PVC/(CaCO,-PMMA-ACR-2)

(b)

Figure 8 SEM micrographs of brittle-fractured surfaces of tensile samples of PVC/nano-CaCO₃ (100/15) composites along the stretching direction (a, apart from the breakpoint; b, close to the breakpoint).

peak of $PVC/(CaCO_3-PMMA-ACR)$ are the biggest among PVC and various $PVC/nano-CaCO_3$ composite. All these indicate that the interfacial interaction between

(a)

PVC and nano-CaCO₃ particles can be improved by modification of nano-CaCO₃ particles surfaces with PMMA and formation of core-shell structured particles.

Morphology and tensile fracture behavior of various PVC/nano-CaCO₃ composites

Morphology of impact samples of PVC/nano-CaCO₃ composites was shown in Figure 7. There are many exposed CaCO₃ particles in PVC/(CaCO₃/ACR) composite. But in PVC/(CaCO₃-ACR2) and PVC/ (CaCO₃–PMMA/ACR) composites the CaCO₃ particles are almost wrapped by organic compounds. All these indicated that ACR was compatible with PVC but not compatible with CaCO₃ particles at the absence of PMMA, so when $(CaCO_3/ACR)$ particles were blended with PVC, ACR was well blended with PVC and CaCO₃ particles were directly dispersed into PVC matrix with quite poor interfacial interaction. When CaCO₃–ACR2 particles were blended with PVC, CaCO₃ particles wrapped with ACR were well dispersed into PVC matrix, so there were almost no exposed CaCO₃ particles in PVC/(CaCO₃-ACR2) composite. After CaCO₃ particles were treated with PMMA, ACR acted as a compatibilizer between CaCO₃-PMMA and PVC and so CaCO₃ particles were well wrapped by polymer matrix. For $PVC/(CaCO_3-$ PMMA–ACR2) composite, there was a good interface between CaCO₃-PMMA-ACR2 particles and PVC, indicating that here ACR was not completely mixed with PVC and acted as a cushion between PVC and CaCO₃, which can result in improvement of the mechanical properties of PVC/CaCO₃ composite.

Figure 8 shows the tensile fracture behaviors of various PVC/nano-CaCO₃ composites. There are many exposed CaCO₃ particles in PVC/(CaCO₃/ ACR) and PVC/(CaCO₃-ACR2) composites both close to and apart from the breakpoint, but the filler particles in $PVC/(CaCO_3 - ACR2)$ composite are apparently more compact with PVC matrix than in PVC/ (CaCO₃/ACR) composite. For PVC/(CaCO₃-PMMA/ ACR) composite, the $CaCO_3$ particles are almost wrapped by organic compounds in impact fractured SEM micrograph but exposed in tensile fractured SEM micrograph, indicating that interfacial interaction of PVC/(CaCO₃-PMMA/ACR) composite was improved but not strong enough. Comparing with PVC/ (CaCO₃–PMMA/ACR) composite, the filler particles in PVC/(CaCO₃-PMMA-ACR-2) composite were well wrapped in PVC matrix apart from the breakpoint and more compact with PVC matrix close to the breakpoint, indicating that interfacial interaction between CaCO₃-PMMA-ACR-2 particles and PVC was strong enough not to be destroyed before a higher critical stress was reached, or stress can be transferred to the filler particles. All these proved that interfacial interaction can be improved by modification of nano-CaCO3 particles surface with PMMA and formation of core-shell structured particles with elastic ACR shell.

CONCLUSIONS

XPS, TEM, and dissolving experiment results indicate that $[(CaCO_3 \text{ particle})/(elastomer)]$ core-shell structured particles can be obtained by vibro-milling and various nano-CaCO₃ particles with different interfacial layers between nano-CaCO₃ particles and PVC can be obtained by the methods used.

The mechanical properties of PVC/CaCO₃ composite were improved if CaCO₃ particles were first treated with PMMA and in succession coated by ACR through vibro-milling, and the calculated results of the interfacial interaction and DMA results reveal that the interfacial interaction between nano-CaCO₃ particles and PVC is the strongest here.

Morphology and tensile fracture behavior studies reveal that to make full use of the synergetic effect of elastomer and rigid particle, the elastomer should not be completely miscible with polymer matrix and there should be strong interfacial interaction between the elastomer and rigid particles.

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